

Production of edible salt from Carlsbad thermal spring water by electro dialysis

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Carlsbad thermal springs

- Springs healing ability discovered by Václav Payer in 1522
- Located in the spa town of Karlovy Vary (Carlsbad), Czech Republic
- Positive effect on digestive tract (liver, stomach, intestine)
- Drinking and bathing cure, wellness procedures
- Drinking cure developed in 18th and 19th century by David Becher, Jean de Carro, Rudolf Mannl, Eduard Hlawacek
- 13 of 79 mineral thermal springs collected and used for drinking cure
- 153,894 spa guests in 2012

Objectives:

- Production of solid crystalline Carlsbader salt for drinking cure at home (follow-up the spa treatment at home)
- Food grade quality salt, easily soluble
- Capacity 20m³ spring water daily (8h shift)
- Concentration technology cheaper than evaporator
- Waste water TDS concentration <2.5g/l (disposal limit, salt loss)



Feed:

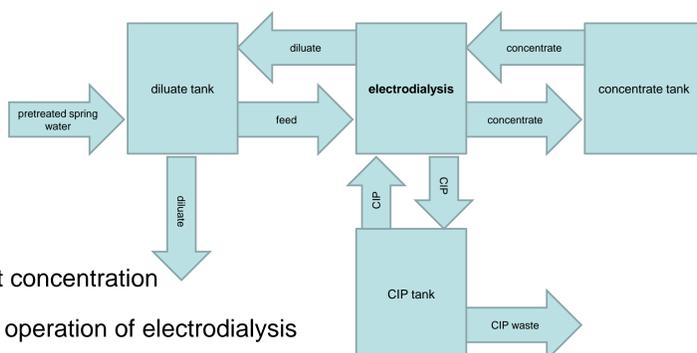
- Carlsbader spring water
- Delivered in 10m³ stainless steel tanks
- Pretreated and cooled down to 40°C

κ [mS/cm]	721	NH ₄ ⁺ [mg/l]	<0.02
pH	8.58	Cl ⁻ [mg/l]	588
TDS [mg/l]	5320	SO ₄ ²⁻ [mg/l]	1610
COD _{Cr} [mgO ₂ /l]	11.8	HCO ₃ ⁻ [mg/l]	2000
Ca ²⁺ [mg/l]	1.51	CO ₃ ²⁻ [mg/l]	30.3
Mg ²⁺ [mg/l]	0.047	F ⁻ [mg/l]	6.28
Ba ²⁺ [mg/l]	<0.0005	NO ₃ ⁻ [mg/l]	<0.5
Sr ²⁺ [mg/l]	<0.001	Total Si [mg/l]	43.4
Na ⁺ [mg/l]	1943	Total P [mg/l]	0.06
K ⁺ [mg/l]	0.56	Total Fe [mg/l]	0.06
Li ⁺ [mg/l]	0.402	Total Mn [mg/l]	0.001

Feed water analysis

ED design by laboratory tests

- ED unit P EDR-Z/10-0.8 with 0.064m² active membrane area (own production)
- Membranes: anion exchange Ralex[®] AMH-PES, cation exchange Ralex[®] CMH-PES
- Estimation of concentrate conductivity limit (salt precipitation)
 - Gradual increase of concentrate conductivity until scaling occurs (visual control of disassembled ED stack), 7 batch tests
- Estimation of ED capacity dependence on diluate conductivity
 - Batch test at concentrate conductivity limit
- Scale-up to industrial size unit (own production)



Technology overview:

- Electro dialysis for salt concentration
- Diluate tank for batch operation of electro dialysis
- Concentrate tank for product storage prior transport to evaporator
- Desalinated water (diluate) discharged to the drain after each batch
- CIP tank for ED chemical cleaning and conservation

ED design:

- ED unit P2 2xEDR-X/100-0.8 with 18.9m² of active membrane area
- Temperature: <40°C controlled by operator (PE limits)
- Voltage: 1.5V/cell
- Circulation flowrate: 5m³/h for both diluate and concentrate
- Safety filter: 100µm (Fe) for both diluate and concentrate
- Manual batch start
- Automatic batch control (batch end, constant concentrate conductivity, voltage and current limits)

ED product:

- Concentrated aqueous salt stream
- NaHCO₃ type
- Detail composition --->
- **19x concentrated**
 - ✓ low volume
 - ✓ low transport costs



Density	kg/m ³	1052
Conductivity	mS/cm	61.8
pH		8.57
TDS (105°C)	mg/l	102000
TS (180°C)	mg/l	62220
TDS (550°C)	mg/l	61400
TSS	mg/l	112
Na ⁺	mg/l	24000
K ⁺	mg/l	1070
Ca ²⁺	mg/l	8.07
Mg ²⁺	mg/l	17.3
Ba ²⁺	mg/l	0.685

Sr ²⁺	mg/l	0.11
SO ₄ ²⁻	mg/l	20500
Cl ⁻	mg/l	7950
F ⁻	mg/l	93.2
HCO ₃ ⁻	mg/l	20400
CO ₃ ²⁻	mg/l	1360
CO ₂ total	mg/l	17600
P total	mg/l	N/A
Fe total	mg/l	0.488
Mn total	mg/l	0.05
CHSK _{Cr}	mg/l	1440

ED product analysis

Final product:

- Evaporate and dry ED product
- Dry solid Carlsbader salt, 3g/package
- To follow-up the drinking cure at home
- Drinking dissolved in tap water at 0,4-5,2% concentration
- Mineral water production at 0,3-1,5g/l
- Additive for food industry (bakery, pastry)



Major ions	g/kg	Trace elements	mg/kg
Na ⁺	314.0	Li ⁺	440
K ⁺	13.7	Ca ²⁺	394
SO ₄ ²⁻	275.0	Si	388
HCO ₃ ⁻	241.0	Mg ²⁺	95.5
Cl ⁻	104.0	P	12.1
CO ₃ ²⁻	42.6	Fe	6.36
		Mn	0.34
		Ba ²⁺	0.30

Final product analysis



Conclusion:

- Industrial ED unit design based on laboratory ED tests
- Scale-up ratio 300:1 still working OK for ED
- Salt produced by the company Original Karlsbader Sprudelsalz since May 2012
- No problems with the technology yet